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## Defect-mediated Jahn-Teller effect in layered LiNiO<sub>2</sub>

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Ni-rich layered lithium transition metal oxide materials, such as  $LiNi_{1-x-y}Mn_xCo_yO_2$  (NMC) and  $LiNi_{1-x-y}Co_xAl_yO_2$  (NCA), are used as cathode materials in lithium-ion batteries [1-4]. Improvements in the energy density and reductions in Co cost can be achieved by increasing the Ni content so that the compositions of these two materials invariably converge toward LiNiO<sub>2</sub> (LNO). Recently, LNO has attracted increased attention for compositional designs of next-generation low-/zero-Co, ultrahigh-Ni layered oxides as alternatives to NMC/NCA. However, the crystallographic and electronic structures of LNO remain subjects of considerable controversy. LNO has been experimentally reported to exist in the  $R\overline{3}m$  phase [5–9] and exhibit semiconductor behavior with a small band gap [10-15], while theoretical studies have reported that stoichiometric  $R\overline{3}m$ LNO is a conductor and unstable [16-22]. Moreover, the Ni ions in LNO are considered as Jahn-Teller (JT)-active Ni<sup>3+</sup> with the electronic configuration of  $t_{2g}^{\phantom{1}6}e_g^{\phantom{1}1}$  [8,10,12,23]. Nevertheless, extended X-ray absorption fine structure and neutron diffraction data have indicated that only local Jahn-Teller distortion (LJTD) exist [8,24-26], implying the absence of long-range collective Jahn-Teller distortion (CJTD), which is the ground state of NaNiO<sub>2</sub> (NNO), a topologically equivalent compound of LNO [23,27,28].

The discrepancy between the experimental and theoretical studies has been explained based on phases other than R3m[22,29–31], building possible orbital ordering patterns [8,21,30,32], defective structures [33-35], time and ordering averages of dynamic stabilization [29,36], and combining theoretical methods [16-20,22,37]; however, few such studies have provided sufficient explanations. The most plausible explanation for this discrepancy was provided by Petit et al. [33]. They established an extra-Ni-based model with a ~1 eV band gap by substituting 25% of Li<sup>+</sup> with Ni<sup>2+</sup> using the self-interactioncorrected local-spin-density method [33]. However, this band gap is extremely large compared with the experimental data of ~0.5 eV, and no LJTD has been reported [11-14]. Moreover, 25% of the Li sites occupied by Ni<sup>2+</sup> appears far-fetched, as (Li<sub>1-x</sub>Ni<sub>x</sub>)NiO<sub>2</sub> has been reported with an x of up to 0.14 [5,13,26,38–40].

In this study, we observed that when Ni/Li antisites are considered, the discrepancy between the experimental and theoretical results of  $R\overline{3}m$  LNO can be solved. Some theoretical and experimental studies have confirmed the existence of Ni/Li antisites in NMC [41–44] and LNO [7,34], but the focus of these

studies was on the removal of Ni/Li antisites and not on the JT effect [45,46]. The defect concentration of  $(Li_{1-x}Ni_x)(Ni_{1-x}Li_x)O_2$  was determined as ~3% by Kanno *et al.* [13]. Based on hybrid density functionals of Heyd-Scuseria-Ernzerhof (HSE06) calculations, LNO with Ni/Li exhibited a band gap of 0.40 eV with LJTD, which agreed with the experimental results. Further discussion on the Kanno's model and our computational details can be found in the Supplementary information (SI) Sect. A (Fig. S1) and B.

We considered a trigonal structure  $(R\overline{3}m)$  comprising layers of NiO<sub>2</sub> slabs, with edge-sharing octahedra, separated by a lithium layer (Fig. 1a). In the ideal structure, Ni ions occupy 3a sites and Li ions occupy 3b sites. Two possible defective LNO types were considered (Fig. 1a): a model with extra Ni (the previous model,  $(\text{Li}_{26}\text{Ni})_{3b}(\text{Ni}_{27})_{3a}\text{O}_{54}$  [5,7,39,40]) and a model with Ni/Li (the model proposed herein, (Li<sub>26</sub>Ni)<sub>3b</sub>(Ni<sub>26</sub>Li)<sub>3a</sub>O<sub>54</sub>). Some theoretical studies have focused on the P21/c phase as it exhibits the lowest energy of LNO among various phases [22,29-31]. However, it should be excluded owing to its inconsistency with experiments.  $P2_1/c$  is significantly different with  $R\overline{3}m$ , as reported by prior experiments [5-9]. Besides, defects were not considered in these studies, which conflicted with the fact that ideal LNO has not yet been synthesized [5,10,13,26,47,48]. The calculated parameters of  $R\overline{3}m$  LNO (a = b = 2.896 Å, c= 14.397 Å) are closer to the experimental data (a = b = 2.883 Å, c = 14.215 Å) [7,49]. Although the magnetism of LNO is affected by many factors, ferromagnetic (FM) is a common approximation [9]. Therefore, we considered LNO as FM for all calculations. Its lattice parameters match with the experimental data, but NiO<sub>6</sub> octahedra in R3m LNO have identical Ni-O bond lengths, contradictory to the varying Ni-O bond lengths reported via experiments [8,24,26]. This discrepancy would be explained based on defective models.

The density of states (DOS) of the ideal  $R\overline{3}m$  LNO indicates that the empty spin-down e<sub>g</sub> band and half occupied spin-up e<sub>g</sub> band ensure the electronic configuration of  $t_{2g}^6 e_g^1$ , corresponding to Ni<sup>3+</sup> (Fig. 1b). The spin-up e<sub>g</sub> band lies on the Fermi level and exhibits a conducting behavior, which is consistent with the previous theoretical studies and contrary to the experimental semiconductor behavior [11–14,16–20]. The extra-Ni model (Li<sub>26</sub>Ni)Ni<sub>27</sub>O<sub>54</sub> also exhibits a conducting behavior. In the antisite model (Li<sub>26</sub>Ni)(Ni<sub>26</sub>Li)O<sub>54</sub>, the band gap of 0.40 eV is consistent with the experimental data (0.2 eV [13] and 0.5 eV [14]). Models with different Ni/Li concentrations also exhibited

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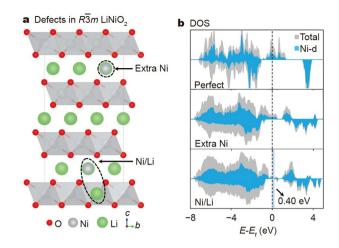
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**Figure 1** (a) Illustration of defects in  $R\overline{3}m$  LNO. (b) DOS of various LNO.

band gaps of ~0.5 eV (Table S1) instead of conducting behaviors. Since the defect concentration was experimentally reported as ~3% [13], we focused on  $(\text{Li}_{26}\text{Ni})(\text{Ni}_{26}\text{Li})\text{O}_{54}$ , whose concentration was 3.7%. The antisite LNO (-333.09 meV f.u.<sup>-1</sup> relative to  $R\overline{3}m$ ) is a thermodynamically stable structure since its energy is lower than that of  $R\overline{3}m$  LNO and close to that of C2/m LNO (-419.38 meV f.u.<sup>-1</sup> relative to  $R\overline{3}m$ ), which is another disputed possibility but exhibits CJTD [22].

Ni<sup>3+</sup> bears the  $^{1}_{2g}$  6eg 1 electronic configuration, which is JT active. Fig. 2 profiles the Jahn-Teller distortion (JTD) of  $R\overline{3}m$  LNO without or with Ni/Li. The ideal LNO shows neither a CJTD nor an LJTD. Chung *et al.* [8] suggested that LNO is in a trimer ordering with three equivalent JTD directions, by neutron diffraction. However, some theoretical and experimental studies suggested that a local or mesoscale trimer structure may exist but the possibility of a long-range trimer ordering should be excluded since it is unstable [22,30,32]. As far as we are aware, although the specific form of distortion is still under debate, no report has denied that JTD occurs locally with three kinds of oriented JT orbitals in LNO. This has been exactly observed in

our antisite LNO. Besides, this is another reason to exclude  $P2_1/c$  LNO, which bears a zigzag JTD [22,29–31]. Only partial NiO<sub>6</sub> octahedra of the antisite LNO distort with three kinds of oriented JT orbitals, and CJTD is absent (Fig. 2b). (Li<sub>26</sub>Ni) (Ni<sub>26</sub>Li)O<sub>54</sub> possesses Q<sub>2</sub> and Q<sub>3</sub> symmetrized octahedral modes (Fig. 2c) [50]. Different bond lengths are grouped as long (~2.08 Å) and short bonds (~1.95 Å), which are in good agreement with the LJTD data obtained by the neutron diffraction (long bond lengths ~2.05 Å, short ones ~1.95 Å) [8].

It is necessary to determine the charge state of Ni ions since only Ni3+ is JT-active. Magnetic moments of nickel ions in (Li<sub>26</sub>Ni)(Ni<sub>26</sub>Li)O<sub>54</sub> are divided into three groups (Fig. 3a), and the projected density of states (PDOS) of Ni ions corresponding to the magnetic moments are illustrated in Fig. 3b. Ni ions with magnetic moments of  $\sim 1.63 \mu_B$  (25.93%),  $\sim 0.85 \mu_B$  (55.56%), and  $0.00 \mu_B$  (18.52%) favor electronic configurations of  $t_{2g}^6 e_g^2$  (Ni<sup>2+</sup>),  $t_{2g}^{6}e_{g}^{1}$  (Ni<sup>3+</sup>), and  $t_{2g}^{6}e_{g}^{0}$  (Ni<sup>4+</sup>), respectively. Ni ions in various charge states contributed to the total charge balance. We observed that Ni3+ ions correspond to JT ions or orbital orderings, while Ni4+ and Ni2+ correspond to charge orderings (Fig. 3a). The competition between orbital ordering and charge ordering has also been demonstrated by Chen et al. [22]. Ni<sup>4+</sup> and Ni<sup>2+</sup> are products of the sharp distortion. The intense interactions between Ni-3d and O-2p orbitals are triggered, resulting in the variation of all Ni-O bond lengths in a NiO<sub>6</sub> octahedron. Distortions in the Q2 and Q3 modes result in the splitting of  $e_g$  bands  $\mbox{[51]}.$  Both  $t_{2g}$  and  $e_g$  bands will split under a JTD, but t<sub>2g</sub> is fully filled and below the Fermi level; therefore, its splitting will not be discussed herein. Every Ni with a 0.85  $\mu_B$ magnetic moment, namely Ni<sup>3+</sup>, exhibits a normal JTD. Though LNO exhibits partial JTD, every NiO<sub>6</sub> octahedron undergoes a distortion, and local orbital ordering is the embodiment of distortions to various extents. Moreover, JTD is not only a characteristic of crystal structures but also the origin of electronic structures, since it causes the splitting of bands and creates band gaps. Besides, such a metal-semiconductor transition along with the orbital ordering in LNO may be related to the spin-Peierls transition, as reported by Hase et al. [52] for

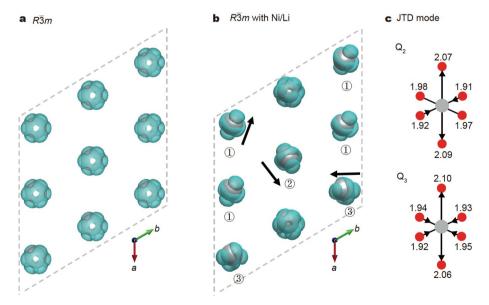


Figure 2 3D spin electron density maps of Ni ions in LNO (a) without and (b) with Ni/Li, respectively. The isovalues for the isosurfaces of the spin densities are 0.05 Å $^{-3}$ . Arrows in (b) imply lengthened Ni–O directions. (c)  $Q_2$  and  $Q_3$  modes of JTD.

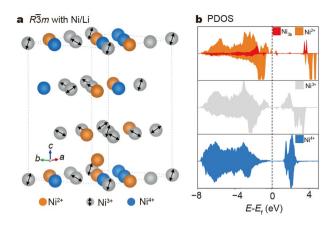


Figure 3 (a) Magnetic moments and (b) PDOS for Ni-d orbitals of corresponding Ni ions for various magnetic moments of LNO with Ni/Li.

## CuGeO<sub>3</sub>.

The Ni/Li antisites consist of Ni<sup>2+</sup> at 3b sites (Ni<sup>2+</sup><sub>3b</sub>) and Li<sup>+</sup> at 3a sites (Li<sup>+</sup><sub>3a</sub>) (Fig. 1a). We compared (Li<sub>26</sub>Ni)(Ni<sub>26</sub>Li)O<sub>54</sub> (with Ni/Li) with the other two models: (Li<sub>26</sub>Ni)Ni<sub>27</sub>O<sub>54</sub> (with only Ni<sup>2+</sup><sub>3b</sub>) and Li<sub>27</sub>(Ni<sub>26</sub>Li)O<sub>54</sub> (with only Li<sup>+</sup><sub>3a</sub>) (Figs S2 and S3). Ni2+3b and Li+3a affect the NiO2 slabs via the size effect and Coulomb interactions. The disturbance from Ni<sup>2+</sup><sub>3b</sub> is weak. Based on the Shannon ionic radius [53], Li<sup>+</sup>, Ni<sup>2+</sup>, and Ni<sup>3+</sup> bear ionic radii of 0.76, 0.69, and 0.56 Å, respectively, and since the ionic radii of Li<sup>+</sup> and Ni<sup>2+</sup> are close, the size effect of Ni<sup>2+</sup><sub>3b</sub> is faint. Moreover, though Ni2+3b carries more charge than Li+3b, the Coulomb interaction can hardly be considered, as interplanar (NiO<sub>6</sub>)<sub>3b</sub> and (NiO<sub>6</sub>)<sub>3a</sub> octahedra are corner-sharing (Fig. S3c) and the Coulomb repulsion between Ni<sup>3+</sup><sub>3a</sub> and Ni<sup>2+</sup><sub>3b</sub> is screened by the shared O ion. Thus, (Li<sub>26</sub>Ni)Ni<sub>27</sub>O<sub>54</sub> has the smallest JTD index (definition in the SI Sect. C) of NiO6 octahedra and exhibits conducting behavior (Figs S2b and S3d). Next, the effects of  $\text{Li}^{+}_{3a}$  in  $R\overline{3}m$  LNO was further confirmed by a pure Li-doping model, Li $_{27}(Ni_{26}Li)O_{54}$ . Li $^{+}_{3a}$  will drastically disturb NiO2 slabs via both O and Ni ions (Fig. 4). Since Li+ is larger than Ni3+, Li+3a disturbs neighboring octahedra via the connected O ions. Then, Ni3+3a witnesses a decrease in the Ni<sup>3+</sup>3a-O bond lengths and an increase in the O-Ni<sup>3+</sup>3a-O bond angle, as the intraplanar NiO6 and LiO6 octahedra are edgesharing. Moreover, since NiO2 layers consist of edge-sharing octahedra, there is considerable Coulomb repulsion between the cations. Li<sup>+</sup> carries less charge than Ni<sup>3+</sup> and the exclusion is weaker; thus, Ni3+3a ions around Li+3a deviate from their original sites. Therefore, Li<sub>27</sub>(Ni<sub>26</sub>Li)O<sub>54</sub> has the largest JTD index (Fig. S3d) and band gap. Li<sup>+</sup><sub>3a</sub> substantially affects the determining factors of the JTD and band gap in  $R\overline{3}m$  LNO.

On the other hand,  ${\rm Li^+}_{3a}$  is rigid and will obstruct the cooperation of JTD. In LNO, NiO<sub>6</sub> octahedra share edges, and thus, they do not independently distort [54]. When Ni/Li are introduced in the ideal C2/m LNO, the rigid  ${\rm Li^+}_{3a}{\rm O}_6$  octahedra reduce the crystal symmetry; therefore, the correlation of individual active Ni<sup>3+</sup> centers is disrupted, and the elongations of Ni<sup>3+</sup>O<sub>6</sub> octahedra do not occur in a uniform direction; rather, three distorted orientations are observed (Figs 2 and 4). The JTD index of NiO<sub>6</sub> octahedra, particularly near the Ni/Li, are also reduced (Fig. S4). Thus, the Ni/Li can trigger the JTD in R3m LNO on one hand, but hamper the further evolution into CJTD (C2/m LNO) on the other hand, by breaking the long-range

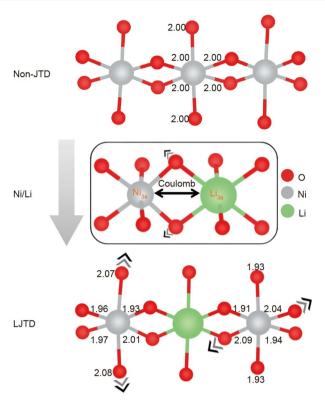


Figure 4 Schematic of the Li<sub>3a</sub> effects.

interaction similar to the interfacial effects in LiMnO $_2$  as reported by Zhu *et al.* [55], ultimately resulting in LJTD with a subtle balance between the charge ordering and orbital ordering. By contrast, unlike Ni/Li antisites in LNO, Ni/Na antisites in NNO are not feasible from the thermodynamic perspective, and the long-range interaction remains, leading to CJTD and a stable C2/m state for NNO (detailed discussions in the SI Sect. D).

To summarize, using hybrid density functional calculations, we demonstrated that Ni/Li antisites are responsible for the local orbital ordering of  $\mathrm{Ni^{3+}}$  in layered LNO, which is in excellent agreement with the experimental data with regards to the LJTD and the small band gap of ~0.5 eV. This can be attributed to the competition between the charge ordering and orbital ordering in LNO, which attains a subtle balance after the Ni/Li antisites are introduced. Therefore, based on the results of this study, we propose that Ni/Li antisites, which have been neglected for a long time in theoretical calculations, deserve more attention in future studies related to the properties of LNO.

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**Conflict of interest** The authors declare that they have no conflict of interest.

**Supplementary information** Experimental details and supporting data are available in the online version of the paper.



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## 层状LiNiO2中的缺陷调制Jahn-Teller效应

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摘要 LiNiO<sub>2</sub> (LNO)基态的晶体结构和电子结构长期以来存在着实验与理论计算不一致的争议.实验上观测到LNO是空间群为 $R\overline{3}m$ 的半导体并且有局部的Jahn-Teller (JT)畸变,但理论计算却表明它是处于亚稳态的金属并且没有任何的JT畸变.本文基于杂化密度泛函理论HSE06,首次模拟了与实验等同浓度(~3%)的Ni/Li反位缺陷对LNO的影响,发现缺陷能够有效调控LNO中的JT效应.在LNO中引入Ni/Li反位缺陷后,其结构发生了局部的JT畸变,并且其带隙值约为0.5 eV,这些计算结果都和实验现象非常吻合.Ni/Li反位通过粒径效应和库伦作用,既能诱发JT畸变,又能阻碍畸变之间的协同作用,避免相变到C2/m,而只产生局部的JT畸变.本文提出了一种新的策略来解释LNO基态晶体结构和电子结构长期以来的争议,对推动富镍层状材料的设计和应用具有重要意义.